



RO-003-1015022

Seat No. _____

B. Sc. (Sem. V) (CBCS) Examination

February - 2019

**BS-IC-502 : Polymer Chemistry & Analytical
Techniques**

Faculty Code : 003

Subject Code : 1015022

Time : $2\frac{1}{2}$ Hours]

[Total Marks : **70**

- Instructions :** (1) Question paper carries total 5 questions.
(2) All the questions are compulsory & carry 14 marks each.
(3) Draw labeled diagram wherever necessary.
(4) Assume suitable data.

- 1 (A) Answer the following questions : 4
(1) _____ is example of natural polymer.
(2) _____ is example of synthetic polymer.
(3) _____ is also called polyamide.
(4) Propylene undergoes _____ polymerization.
- (B) Answer in brief : (Any **one** out of two) 2
(1) What is degree of polymerization?
(2) Define: Polymer.
- (C) Answer in detail : (Any **one** out of two) 3
(1) Describe crystallinity in polymer with processes.
(2) Explain abrasive resistance and coefficient of friction in detail.
- (D) Write a note on : (Any **one** out of two) 5
(1) Explain glass transition temperature in detail.
(2) Explain weight average and number average molecular weight in detail.

- 2 (A) Answer the following questions : 4
- (1) In mechanism of addition polymerization there are _____ stages.
 - (2) _____ is used as catalyst in addition polymerization.
 - (3) Give full form of PVC.
 - (4) _____ is used as coloring agent for polymer.
- (B) Answer in brief : (Any **one** out of two) 2
- (1) Define: Addition polymerization
 - (2) Define: Condensation polymerization
- (C) Answer in detail : (Any **one** out of two) 3
- (1) Explain condensation mechanism for polymerization.
 - (2) Explain compounding of polymer in detail.
- (D) Write a note on : (Any **one** out of two) 5
- (1) Explain compression molding process with diagram.
 - (2) Write a note on blow molding process.
- 3 (A) Answer the following questions : 4
- (1) Give structure of bisphenol A.
 - (2) Give structure of polycarbonate.
 - (3) Write uses of polythene.
 - (4) Sulfur is used in _____ process for semisynthetic rubber manufacturing.
- (B) Answer in brief : (Any **one** out of two) 2
- (1) Write types of isoprene polymer.
 - (2) Explain urea formaldehyde polymer in detail.
- (C) Answer in detail : (Any **one** out of two) 3
- (1) Explain manufacturing of epoxy resin in detail.
 - (2) Explain manufacturing of Nylon-6,6 in detail.

- (D) Write a note on : (Any **one** out of two) **5**
- (1) Explain addition polymerization mechanism, use and properties of Polythene.
 - (2) Explain addition polymerization mechanism, use and properties of Polystyrene.
- 4 (A) Answer the following questions : **4**
- (1) Potentiometric titration method can be used for coloured solution. True/False?
 - (2) Enlist types of conductance.
 - (3) _____ can be used as detector for most of the Polarimeters.
 - (4) Enlist various light sources used in colorimetric analysis.
- (B) Answer in brief : (Any **one** out of two) **2**
- (1) Write principle of conductometric titration.
 - (2) Discuss any one type of Potentiometric titration.
- (C) Answer in detail : (Any **one** out of two) **3**
- (1) Draw only diagram of Polarimeter.
 - (2) Write advantages of Potentiometric titration.
- (D) Write a note on : (Any **one** out of two) **5**
- (1) Discuss Refractometry method with diagram.
 - (2) Explain colorimetric method in detail.
- 5 (A) Answer the following questions : **4**
- (1) _____ containers are used for the sampling gases like oxygen, nitrogen, methane etc.
 - (2) Give full form of TCD.
 - (3) NMR spectroscopy is used to detect functional group present in hydrocarbon. True/False?
 - (4) The range of light source for UV spectroscopy is _____ nm to _____ nm.

- (B) Answer in brief : (Any **one** out of two) **2**
- (1) Write principle of chromatography.
 - (2) Draw only diagram of atomic emission detector.
- (C) Answer in detail : (Any **one** out of two) **3**
- (1) Discuss basic sampling rules for sampling technique.
 - (2) Explain flame ionization detector with diagram.
- (D) Write a note on : (Any **one** out of two) **5**
- (1) Enlist and explain classification of chromatographic method.
 - (2) Discuss IR spectroscopy with neat diagram.
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